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A Novel Method for the Preparation of 4-Arylimidazolones

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ABSTRACT

A series of 4-arylimidazolones have been accessed *via* late-stage, palladium-mediated arylation of acetone- and cyclohexanone-derived 4-chloroimidazolones. The 4-chloroimidazolones were prepared *via* a novel rearrangement of the corresponding imidazolone *N*-oxides. This communication serves as an expansion of chemistry originally developed for our glucagon receptor antagonist program.

The imidazolone moiety is a structural motif found in the Kottamide family of natural products and glycine transporter type 1 inhibitors such as GSK 2137305.^{1,2} In our laboratories, we have utilized the imidazolone heterocycle in the preparation of a series of orally available glucagon receptor antagonists for the potential treatment of type II diabetes mellitus (T2DM).³

Our initial approach to the synthesis of 4-aryl-substituted spiroimidazolones (Scheme 1) involved starting with an *N*-Boc-arylglycine (1), coupling with a primary amine to afford amide 2, and deprotection resulting in amine 3. Condensation of 3 with a ketone afforded 4, which upon oxidation with NBS or *t*-BuOCl afforded the desired 4-arylimidazolone 5.

For our glucagon receptor antagonist program, it was desired to develop an effective means to install 4-aryl and 4-heteroaryl groups at a late stage in the synthetic sequence, rather than at the first step, as was the case with the *N*-Boc-arylglycine approach.

Previous reports had described the synthesis of imidazolone N-oxides for use as a chiral synthon in the case of compounds $\bf 6$ and $\bf 7a-c$ (Figure 1) and as an example of unique chemical matter ($\bf 8a$). More recently, imidazolone N-oxides $\bf 8b$ and $\bf 9$ were used to prepare a series of structurally unique 4-alkenylimidazolones and 6-acyl-3-benzyl-2,2-dialkyl-1,3-diazabicyclo[3.1.0]hexan-4-ones. 9

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Direct arylation of the 4-position of imidazolone *N*-oxides has also been demonstrated. ¹⁰

Scheme 1. Initial Synthetic Approach to 4-Arylimidazolones

Figure 1. Proposed synthetic approach to 4-arylimidazolones.

For our purposes, we hypothesized that an imidazolone *N*-oxide such as **10** could be transformed into the corresponding 4-haloimidazolones **11a** and **11b** in a manner analogous to the conversion of pyridine *N*-oxides to 2-halopyridines. Palladium catalyzed cross-coupling of **11a** or **11b** with the requisite boronic acid or ester would afford the desired 4-arylimidazolone **5**. This approach was originally validated in our previously described glucagon receptor antagonist efforts.³

With this initial success, we decided to further probe the generality of this chemistry. The results of our efforts are described herein.

Glycinamide 12 was deprotected with TFA to afford amine 13 (Scheme 2). Heating of 13 in refluxing acetone in

Scheme 2. Preparation of 4-Chloroimidazolone 16

the presence of 3 Å molecular sieves afforded the dihydroimidazolone **14**. Oxidation of **14** with 2.2 equiv of *m*-CPBA afforded the nitrone **15**. Treatment of **15** with POCl₃ in the presence of Hunig's base in refluxing toluene, followed by quenching of the reaction with brine and silica gel chromatography (run 1), afforded the chloroimidazolone **16** in 56% yield. Conversely, quenching of the reaction with saturated aqueous sodium bicarbonate followed by silica gel chromatography (run 2) afforded none of **16**, but rather a 56% yield of the hydrolysis product, hydroxyimidazolone **17**.

Additionally, dihydroimidazolone **14** can be converted to chloroimidazolone **16** *via* a second route. Treatment of **14** with *tert*-butyl hypochlorite resulted in *N*-chlorination of the dihydroimidazolone. Subsequent treatment with triethylamine resulted in elimination to the imidazolone **18**. Interestingly, treatment of **18** with *m*-CPBA did not provide the nitrone **15**, but rather the 4-hydroxyimidazolone **17**. ¹³

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Table 1. Arylation of Chloroimidazolone 7^a

entry	Ar	yield
Citity	Al	yiciu
19a	72	42%
19Ь	F	45%
19c	₹ F	57%
19d	F	47%
19e	Me	44%
19f	- Za Me	60%
19g	Me	61%
19h	OMe	47%
19i	ر OMe	24%
19j	MeO	7%
19k	3. N	49%
191	N N	45%
^a Isolated yields.		

As was seen with the treatment of nitrone 15, treatment of 17 with POCl₃ and Hunig's base in toluene at reflux also afforded the chloroimidazolone 16.

To avoid hydrolysis of the chloroimidazolone on work-up, the conversion of 15 to 16 was immediately followed by removal of the volatile reactants *in vacuo* and Suzuki coupling with the requisite boronic acid or boronic ester to afford 4-arylimidazolones 19a—I (Table 1). In general, the Suzuki coupling products were isolated in modest yields with a variety of substituted aromatics. Suzuki products arising from coupling with methoxy-substituted phenyl boronic acids were isolated in the lowest yields, with only a minimal amount of the *ortho*-methoxyphenyl imidazolone 19j being isolated. Pyridinylimidazolones 19k and 19l were isolated in yields similar to those observed with phenylimidazolones.

At this point, we explored the palladium-mediated arylation of cyclohexanone-derived 4-chlorospiroimidazolones (Table 2). The preparation of imidazolone *N*-oxide **20** and 4-chloroimidazolone **21** proceeded in a manner similar to that observed with the preparation of the analogous

Table 2. Preparation of 4-Aryl Spirocyclohexylimidazolones^a

^a Isolated yields.

221

22m

22n

22o

acetone-derived intermediates **15** and **16**. With the exception of the *N*-methylpyrazole compound **220**, moderate isolated yields were observed across the board, including the *o*-methoxyphenyl compound **22k**.

41%

45%

43%

17%

The imidazolone motif has seen a significant increase in utility in recent years. A wide variety of unique synthetic transformations have emerged from imidazolone *N*-oxides in particular. In our laboratories, we have demonstrated a novel method for the synthesis of 4-chloroimidazolones, and their subsequent palladium-catalyzed cross-coupling with arylboronic acids. This approach allows for the rapid, late-stage incorporation of a wide variety of aryl and heteroaryl groups that would not be readily available *via* the original arylglycine-based approach.

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Supporting Information Available. Experimental procedures and analytical data for all new compounds. This material is available free of charge via the Internet at http://pubs.acs.org.

The authors declare no competing financial interest.

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